

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Applicant : Natarajan et al.                      Art Unit : 1762  
Serial No. : 10/676,434                      Examiner : Michael Cleveland  
Filed : September 30, 2003                      Conf. No. : 7054  
Title : SOLVENT MIXTURES FOR AN ORGANIC ELECTRONIC DEVICE

**MAIL STOP AF**

Commissioner for Patents  
P.O. Box 1450  
Alexandria, VA 22313-1450

REPLY TO ACTION OF FEBRUARY 7, 2007

In reply to the Office Action of February 7, 2007, Applicant submits the following remarks. Applicant respectfully requests reconsideration in view of these remarks.

Lyon

Claims 1-5, 8-9, 15-16, 28-33, 35, 38-39 and 51 were rejected under 35 U.S.C. § 103(a) as being unpatentable over WO 02/069119 ("Lyon"). The applicant respectfully disagrees.

Claim 1 is directed to a method comprising depositing an organic polymer solution on the first electrode, wherein the solution includes a first solvent, at least one organic polymer, a second solvent and a third solvent, the first solvent has a high solubility and a faster evaporation rate than the second solvent, the second solvent has a very low solubility, and the third solvent has a surface tension less than 30 dynes/cm and is less than about twenty weight percent of the solution. Claim 28 is directed to a method of forming an organic polymer layer comprising mixing a first solvent, second solvent and third solvent with an organic polymer. The three solvents have the characteristics described above.

The Examiner argues that "it would have been obvious to one of ordinary skill in the art at the time of invention to have used 20 weight percent of xylene" (Office Action, Page 3). The Examiner relies on Lyon's teaching that a first solvent can be a mixture of two different solvents (page 3). However, the Examiner is confusing Lyon's first and second solvents with the first and second solvents of applicant's claims. Lyon describes the first solvent as having a relatively high boiling point (Lyon, page 3). A solvent with a high boiling point would have a slower evaporation rate when compared to a solvent with a lower boiling point. Thus, Lyon suggests

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